

Original Research Article

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Effect of Blending of Fats and Oils on Physico-chemical Properties of Ghee

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ABSTRACT

Keywords

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The changes in the ghee physico-chemical properties due to blending with different fats and oils at different levels were studied. Ghee was blended with Groundnut oil, Coconut oil, Hydrogenated fat and animal body fat at 10% and 15% respectively. The blended ghee samples were analysed for Reichert–Meissl, Polenske, saponification, iodine values and butyro refractometer reading at 40°C. The study revealed that an increase in B.R. reading and Iodine value are due to blending with groundnut oil and hydrogenated fat, also an increase in the polenske and saponification values are due to blending with coconut oil where as decrease in the B.R. reading and Iodine value are due to blending with coconut oil and decrease in the Reichert–Meissl, Polenske, saponification values are due to blending with hydrogenated fat and animal body fat at 10 & 15 per cent respectively.

Introduction

Ghee is well known for characteristic flavor aroma and taste. It is widely used in the preparations of culinary purpose and also in the preparation of sweets occupying important place in Indian diets. India produces approximately 5.34 million tonnes of ghee annually (BAHFS, 2015). It has got unique fatty acid profile ranging from short chain, long chain, saturated and unsaturated fatty acids besides having fat soluble vitamins A, D, E and K. It is a premium fat rich dairy product, attaining commercial significance to dairy industry.

Because of the seasonal variation the supply of milk and ghee falls short of demand resulting in the producers or the middle-men involved in the ghee trade, tend to adulterate ghee with cheaper oils and fats like vegetable oils, animal body fats, hydrogenated fats and sometimes even the non-edible mineral oils, especially during lean season to earn more money. Thus the problem of adulteration has assumed a very serious dimension.

Milk fat, like other fats and oils, is characterized by certain physicochemical properties i.e., Reichert–Meissl (RM) value, Poleske value, saponification value, iodine value, butyro-refractometer (BR) reading and

free fatty acids(FFA)which have been found to be the basis for the fixation of certain physicochemical constants for defining the chemical quality of the product. These constants serve as an indication of the types of component fatty acids present in fats. They also enable the detection of fat adulteration qualitatively and, in some instances, quantitatively.

These physicochemical properties, however, exhibit some natural variation depending upon factors such as method of manufacture, age and condition of the sample, species, breed, individuality of animal, stage of lactation, number of lactation (age of animal), season of the year, region of the country, feed of the animal (Doctor *et al.*, 1940; Aneja *et al.*, 2002; De 2005).

In order to ensure a genuine pure safer quality product to the consumer, the government of India has prescribed the compositional standards for ghee, under FSSAI and AGMARK rules. FSSAI (2011) has prescribed the requirements for two physicochemical constants, namely B.R. reading at 40 °C and RM value, plus FFA (as oleic acid) and moisture for defining the quality of milk fat in terms of ghee/anhydrous milk fat/anhydrous butter fat. While, AGMARK standards (AGMARK, 1981) specify the physicochemical constant, that is B.R. reading, RM value and Polenske value, besides FFA and moisture for the same purpose. The other three physicochemical constants such as iodine value, saponification value and Polenske value for milk fat range between 26 and 35, 210 and 233 and 1.2 and 2.4, respectively (Jeness and Patton 1969).

In the present article different market, desi and control ghee samples were blended with different fats and oils at two different percent (10, 15) and have checked for their effect on physico-chemical properties of ghee samples.

Materials and Methods

Chemicals and reagents

All the chemicals used during the investigation were AR grade and procured from standard suppliers (HI MEDIA®).

Preparation of ghee sample

Fresh cream was procured from student experimental dairy plant (SEDP) of Dairy Science College, Karnataka Veterinary Animal and Fisheries Sciences University (KVAFSU), Bangalore and used for the ghee preparation by the method given by De (2005). Some of the commercial branded as well as desi ghee samples which were procured from local market of Bangalore used to assess the physico-chemical quality and the purity.

Ghee samples

Samples of ghee from organized sector (samples A, B, C, D and E) and unorganized sector (samples F, G, H and I) were collected from local market of Bangalore.

Blending of ghee samples with fats and oils

For the blending of ghee samples, with blending agents like groundnut oil, coconut oil, hydrogenated fat and animal body fat, the blending agents were heated and maintained at 65-70°C for 10 min before blending, then they were blended to ghee samples individually at 10 and 15 per cent to study their impact on physico-chemical constants of ghee.

Physico-chemical properties of ghee samples

The ghee samples which were procured from the market, blended ghee samples with fats

and oils and also the ghee which was prepared in the SEDP were analyzed for different physico-chemical constants like RM value, Polenske value, saponification value, iodine value and B.R. reading at 40°C to know their effect on physico-chemical properties of ghee by [SP: 18 (Part XI) – 1981].

Statistical analysis

The data was analyzed using R software {R Programme, R-Version 3.1.3(2015-3-09), Copyright © 2015} both one way and two way Completely Randomed Design (CRD) which is the most appropriate for the study.

Results and Discussion

Effect of blending agents on physico-chemical constants of ghee

In the present study different fats and oils were blended at varied levels to check the variation in the physico-chemical constants of ghee.

B. R. reading

Butyro-Refractometer reading increases with the increase in blending of groundnut oil. The control, market, and desi ghee samples scored a B.R. reading of 43, 44.5 and 44.5 at 0% adulteration. When these ghee samples were blended with groundnut oil at 15% the value was increased to 44, 44.5, and 45 respectively. Upadhyaya (2014) also reported that, there is an increase in the B.R. reading when the ghee samples were adulterated with groundnut oil.

From the (table 1) it is seen that the B.R. reading decreases with an increase in blending with coconut oil. When ghee was adulterated with coconut oil at 10% level, the B.R. reading was 42.5. When the level of adulteration was increased to 15% the value

of B.R. reading was decreased to 42.0. The control ghee showed an average B.R. reading of 43.0. Similar results were seen in case of desi and market ghee samples. The decrease in the B.R. reading of ghee when it is adulterated with coconut oil is mainly due to the presence of short chain fatty acids like lauric, palmitic acids in higher amount. Zachariah (2010) also reported that, there is a decrease in BR of milk fat when the ghee was blended with coconut oil at 10, 20 per cent. When the ghee samples were blended with hydrogenated fat and animal body fat at 10, 15% level showed an increase in the B.R. reading value (table 1).

Iodine value

Iodine value is a measure of the extent of unsaturation of fat. The value varies with the type and proportion of unsaturated fatty acids present in a fat. Ghee contains about 35% of total unsaturated acids, most of which consists of oleic acid; the IV of ghee lies around 30.

In the present investigation the iodine value of control, market and desi ghee samples recorded the values of 34.5, 33.79, and 34.76, respectively. The blending of GO, CO, HF significantly altered the iodine value, while the recorded adulteration with respect to ABF blending found to be non-significant.

The blending of GO and HF at 15per cent level had significantly increased the iodine value from 34.50 to 46.21 and 43.18 respectively in control ghee. When the same control ghee is blended with CO at 15 per cent level the value decreased to 28.22. By blending of CO there is a decrease in the iodine value is observed it is mainly due to coconut oil is having an Iodine Value of around 8-9, lower than ghee. So, lower Iodine Value can be observed in ghee by adulteration with Coconut oil (Bolton, 1999).

Reichert-Meissl value

RM value is the measure of steam volatile water soluble fatty acids (butyric acid and caproic acid). In the present study the RM value of control, market and desi ghee samples recorded the value of 26.8, 23.36 and 23.83 respectively.

There is a decreasing effect on RM value, with an increase in blending of GO, CO, HF and ABF in control, market and desi ghee samples (Table.3). Control ghee showed a values of 23.38, 23.95, 23.33 and 23.38 when blended with GO, CO, HF and ABF at 15 per cent respectively.

Table.1 Studies on the impact of adulterants on B.R. reading

Ghee samples	BR	GO		CO		HF		ABF		CD (P<0.05)
	0%	10%	15%	10%	15%	10%	15%	10%	15%	
Control	43.0 ^a	43.50 ^b	44.0 ^b	42.50 ^b	42.0 ^b	44.0 ^b	44.50 ^b	43.0 ^a	43.50 ^a	0.40
Market	44.50 ^a	45.0 ^b	45.50 ^b	44.0 ^b	43.50 ^b	44.50 ^b	45.0 ^b	45.0 ^b	45.50 ^b	0.38
Desi	44.50 ^a	45.0 ^b	45.50 ^b	44.0 ^b	43.50 ^b	45.0 ^b	45.50 ^b	45.0 ^b	45.50 ^b	0.32

Table.2 Studies on the impact of adulterants on Iodine value

Ghee samples	IV	G O		CO		HF		ABF		CD (P<0.05)
	0%	10%	15%	10%	15%	10%	15%	10%	15%	
Control	34.50 ^a	43.19 ^b	46.21 ^b	30.20 ^b	28.22 ^b	40.65 ^b	43.18 ^b	35.32 ^a	36.44 ^a	2.8
Market	33.79 ^a	42.62 ^b	45.71 ^b	29.82 ^b	27.83 ^b	37.92 ^b	39.92 ^b	34.90 ^a	35.20 ^a	2.66
Desi	34.76 ^a	43.61 ^b	46.66 ^b	30.78 ^b	28.79 ^b	40.89 ^b	43.99 ^b	35.92 ^a	36.72 ^b	2.7

Note:

- All the results are average of three trials.
- Similar superscripts indicate non – significance at the corresponding critical difference
- Statistical analysis was made only with the control ghee sample.

GO= groundnut oil, CO= coconut oil, HF= hydrogenated fat, ABF= animal body fat

Table.3 Studies on the impact of adulterants on R M value

Ghee samples	RM value	GO		CO		HF		ABF		CD (P<0.05)
	0%	10%	15%	10%	15%	10%	15%	10%	15%	
Control	26.80 ^a	24.40 ^b	23.38 ^b	24.81 ^b	23.95 ^b	24.38 ^b	23.33 ^b	24.38 ^b	23.38 ^b	0.09
Market	23.36 ^a	21.28 ^b	20.37 ^b	21.69 ^b	20.96 ^b	21.25 ^b	22.33 ^b	21.26 ^b	20.34 ^b	0.18
Desi	23.83 ^a	21.70 ^b	20.78 ^b	22.11 ^b	21.37 ^b	21.68 ^b	20.74 ^b	21.69 ^b	20.74 ^b	0.2

Table.4 Studies on the impact of adulterants on Polenske value

Ghee Samples	Polenske Value	GO		CO		HO		ABF		CD (P<0.05)
	0%	10%	15%	10%	15%	10%	15%	10%	15%	
Control	1.21 ^a	1.11 ^a	1.06 ^a	2.10 ^b	2.48 ^b	1.14 ^a	1.08 ^a	1.14 ^a	1.07 ^a	0.28
Market	1.46 ^a	1.34 ^a	1.29 ^a	2.32 ^b	2.70 ^b	1.36 ^a	1.31 ^a	1.38 ^a	1.33 ^a	0.40
Desi	1.23 ^a	1.13 ^a	1.18 ^a	2.12 ^b	2.50 ^b	1.16 ^a	1.10 ^a	1.16 ^a	1.11 ^a	0.35

Table.5 Studies on the impact of the adulterants on Saponification value

Samples fats/ oils	Saponificat ion Value	G O		CO		HO		ABF		CD (P<0. 05)
		0%	10%	15%	10%	15%	10%	15%	10%	
Control ghee	230.66 ^a	226.17 ^b	224.52 ^b	231.72 ^a	231.92 ^a	226.19 ^b	224.32 ^b	226.20 ^b	224.81 ^b	1.85
Market ghee	233.0 ^a	228.62 ^b	226.14 ^b	234.64 ^a	234.82 ^a	229.11 ^b	227.20 ^b	228.86 ^b	224.21 ^b	1.42
Desi ghee	232.0 ^a	228.15 ^b	226.68 ^b	231.22 ^a	231.84 ^a	228.48 ^b	226.76 ^b	226.50 ^b	224.33 ^b	2.23

Note:

- All the results are average of three trials.
- Similar superscripts indicate non – significance at the corresponding critical difference
- Statistical analysis was made only with the control ghee sample.

GO= groundnut oil, CO= coconut oil, HF= hydrogenated fat, ABF= animal body fat

The decreasing effect of these adulterants is mainly due to the absence of the C₄ fatty acid (butyric acid) in the adulterants. Other ghee samples like market and desi ghee samples also showed a decrease in the values when blended with the GO, ABF and HF adulterants.

Polenske value

Polenske value is the measure of steam volatile water insoluble fatty acids (caproic and caprylic). In the present investigation the control, market and desi branded ghee samples recorded a polenske value of 1.21, 1.46 and 1.23 respectively.

In the present investigation it was observed that (table 4) there was a decrease in the polenske value of control, market and desi ghee samples when they had blended with GO, HO and ABF at 10 and 15 per cent. But there was an increase in the polenske value when these ghee samples were blended with coconut oil at both the level of adulteration. The control ghee had increased its value from 1.21 to 2.1, and 2.48 at 10 per cent and 15 per cent level of adulteration with coconut oil respectively. The increase in the polenske value in the coconut oil adulterated ghee is mainly due to the presence of higher amount of short chain fatty acids in coconut oil. The results are in agreement with the results given

by Upadhyaya, (2014)

Saponification value

Saponification value is the measure of average molecular weight of fatty acids present in the milk fat. In the present study the control, market and desi branded ghee samples recorded the saponification value of 230.66, 233.0 and 232 respectively.

From the (table 5) it is observed that there was a decrease in the value is observed in all the three ghee samples by adulterating them with the adulterants like GO, HF and ABF. There was a slight increase in the value is observed in the CO adulterated control ghee from 230.66 to 231.92 at 15 per cent adulteration. The results obtained are in accordance with the results given by Upadhyaya, (2014)

In conclusion the results of the present investigation showed that adulteration of ghee with different fats and oils resulted in changes in physico-chemical characteristics and some of the changes could be utilized in detecting adulteration of ghee with coconut oil, groundnut oil, hydrogenated fat and animal body fat. However, the changes in values of several physicochemical characteristics as a result of adulteration were not sensitive enough to detect adulteration and may be an

indication for adulteration having fair knowledge of impact of blending the adulterants on physicochemical characteristics was essential.

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